Organic reactions: classify the change

Teacher / technician

YEAR 11 / LESSON 72 / CHEMISTRY + SCIENCE (DOUBLE AWARD)

Staff preparation and answer guidance. Keep separate from student sheets. Written and model-based lesson; no laboratory procedure is implied.

Coverage and delivery

Shared core: 4.6 with supporting alkane/alkene examples. No reaction mechanisms required.

Allow about 60 minutes. Follow the student sequence and use the answers to check reasoning.

Requisition • class of 30, working in 15 pairs

Item	Per pair / setup	Class total
Booklet / writing tools	1 per learner	30 sets
Model kit / periodic table	1 set per pair	15 sets

Immediate safety

Desk activity only; keep small model pieces away from mouths.

Preparation and delivery

Read alongside the student method

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1. Provide 30 booklets and 15 model kits or reusable atom counters; no chemicals are needed.

2. Use 10 minutes retrieval, 25 independent/pair tasks, 15 explanation and 10 corrections. Keep first attempts visible.

3. Use counters instead of chemicals. Chlorine, bromine and ignition equipment are deliberately absent from this lesson.

4. Clarify that the mono-substitution equation is the restricted model; actual halogenation can produce mixtures.

5. Check all carbon valencies, formula subscripts and equation coefficients. Label any supplied numbers as practice data, not class observations.

Close of lesson

• Return shared equipment and retain corrected reasoning.

Check before teaching

Check the cards, reference data and quantities against this requisition. Keep the staff answer guide separate until the checking stage. Supplied numerical practice datasets are labelled examples; they are not class measurements.

Answers and teaching notes

Expected explanations • do not give as experimental observations

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- Substitution; chloromethane and hydrogen chloride.
- Addition; 1,2-dibromoethane.
- $2C_2H_6 + 7O_2 \rightarrow 4CO_2 + 6H_2O$.
- Its double bond becomes a single bond; one Br atom attaches to each former double-bond carbon.
- The displaced H combines with the second Cl; atom conservation requires both products.
- No. Consider the actual rapid burning/oxidation process and products; oxygen can participate in other reactions.

References and provenance

Pearson Chemistry specification

Prepared 22 November 2026. Original wording, tables and questions. Sources inform the concepts and technique; the lesson scope and classroom quantities are explicitly described here. No laboratory trial or final school risk approval is claimed.